



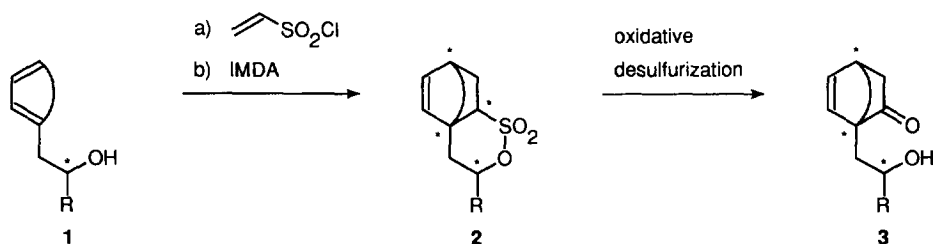
## Desulfurization of Sulfones with Simultaneous Methylenation

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**Abstract:** Desulfurization of the  $\delta$ -sulfones **2** with simultaneous formation of an exocyclic olefin is achieved by alkylation of **2** with (iodomethyl)trimethylsilane and subsequent treatment of the resultant products **7** with tetra-*n*-butylammonium fluoride. Application of this two-step procedure to sulfone **8** leads to 1,3-diene **10**, an intermediate for the synthesis of the highly oxygenated 1,10-*secodesmanolides* eriolanin and eriolangin. Copyright © 1996 Elsevier Science Ltd

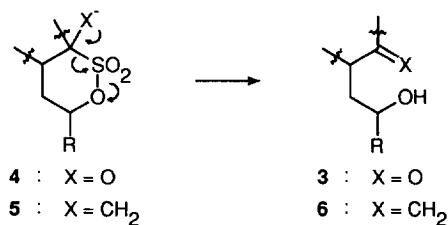
We recently reported a method for the desulfurization of sulfones<sup>1</sup> to hydroxy ketones *via* borylation/peracid oxidation. Coupled with the preparation of  $\delta$ -sulfones **2** by intramolecular Diels-Alder reaction of the vinylsulfonates derived from hydroxyalkyl substituted cycloalka-1,3-dienes **1**, this procedure allows for a formal [4+2] cycloaddition of ketene to **1** in a completely regioselective and highly diastereoselective fashion (Scheme 1).<sup>2</sup>



Scheme 1. Formal [4+2] cycloaddition of ketene to cycloalka-1,3-dienes **1**

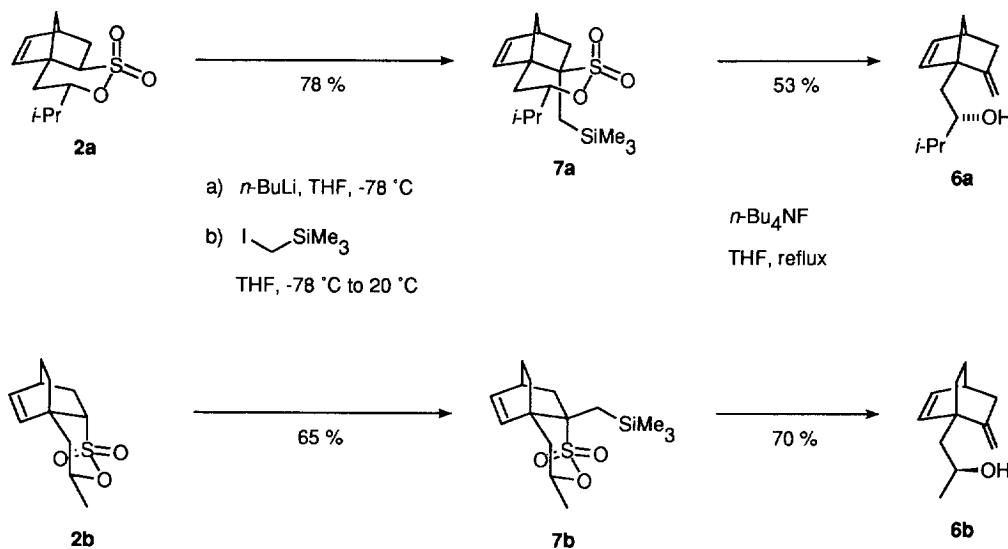
Since the  $\alpha$ -oxygenated intermediates **4** involved in the oxidative cleavage smoothly decomposed in the depicted sense, we reasoned that the carbanions **5**<sup>3</sup> would similarly break down to give the bishomoallylic alcohols **6** *via*  $\beta$ -elimination (Scheme 2).

While a carbonyl methylenation of **3** to **6** would be a feasible alternative for simple substrates, a desulfurization with simultaneous formation of an olefin streamlines the route to **6** and, more importantly, such a process is even applicable if an oxidative sulfone cleavage would lead to a base sensitive ketone, which might undergo side reactions during olefination.

Scheme 2. Sultone cleavage *via* elimination

Using **2b**<sup>2</sup> as a model substrate, we first attempted a one-pot conversion to dienol **6b** by alkylation of the lithiated<sup>2</sup> sultone with chloromethylmagnesium chloride.<sup>4</sup> However, in contrast to the clean olefination of lithiated sulfones with electrophilic monohalogenocarbonoids,<sup>5</sup> only a complex product mixture was obtained from **2b**.

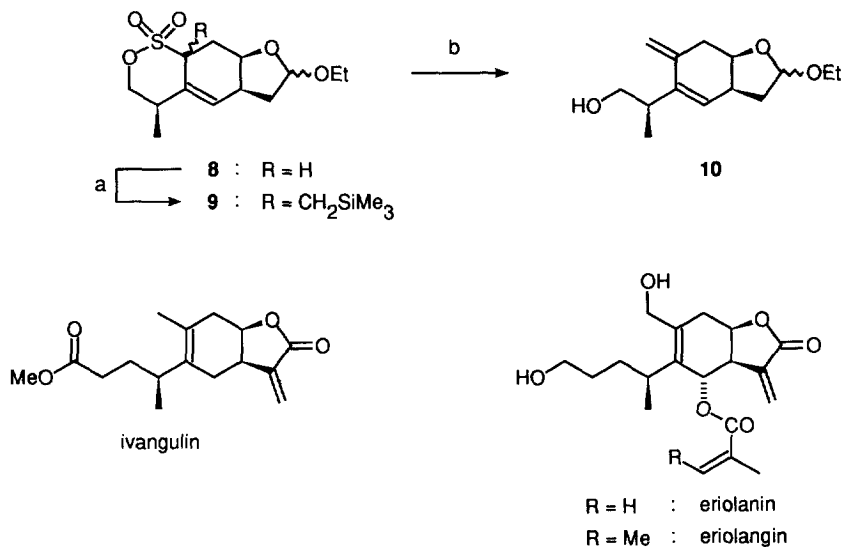
Gratifyingly, the desired transformation of **2a,b** to **6a,b**<sup>6</sup> was achieved in good yield by alkylation of the sultones with (iodomethyl)trimethylsilane<sup>7,8</sup> followed by fluoride-induced elimination<sup>9,10</sup> of the resultant silanes **7a,b**<sup>11</sup> using tetra-*n*-butylammonium fluoride (Scheme 3). Since the dienols **6a,b** are formal [4+2] adducts of allene with the hydroxyalkyl dienes from which the sultones **2a,b** were prepared,<sup>2</sup> vinylsulfonyl chloride<sup>12</sup> can serve as a regio- and stereoselectively reacting allene equivalent for the intramolecular Diels-Alder cycloaddition, too.

Scheme 3. Desulfurization of sultones **2** to dienols **6**

After some minor modifications, the two-step procedure described above was successfully applied to sultone **8**<sup>13,14</sup> as well (Scheme 4). As we already noted during our synthesis of the 1,10-*seco*-eudesmanolide ivangulin *via* methylation of **8**,<sup>14</sup> the use of methyl lithium is essential for complete lithiation. Moreover, efficient alkylation of the resultant allyllithium species required addition of (iodomethyl)trimethylsilane at

room temperature and subsequent warming to 40 °C for 30 min. Fluoride-induced elimination of silane **9** proceeded uneventfully within 1 h to give the acid sensitive dienol **10** as a mixture of two acetal epimers.

Both termini of the 1,3-diene unit in **10** are activated towards an oxygenation<sup>15</sup> and thus, this compound represents a promising intermediate for the total synthesis of the highly oxygenated, antileukemic<sup>16</sup> 1,10-*secocudesmanolides* eriolanin<sup>17-19</sup> and eriolangin<sup>17</sup> by a route similar to the one that led to ivangulin.<sup>14</sup>



Scheme 4. Preparation of 1,3-diene **10**, an intermediate for the synthesis of eriolanin and eriolangin. a: (i) MeLi, THF, -78 °C to 20 °C, (ii) ICH<sub>2</sub>SiMe<sub>3</sub>, 20 °C to 40 °C, 71 %; b: *n*-Bu<sub>4</sub>NF, THF, reflux, 65 %

In summary, a novel method for the desulfurization of sultones has been established that complements our procedures for oxidative<sup>2</sup> and reductive<sup>20</sup> sultone cleavage. Elaboration of the epimeric acetals **10** to eriolanin and eriolangin is currently under investigation in our laboratories.

#### Acknowledgment

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#### REFERENCES AND NOTES

- For reviews on sultone chemistry, see: Buglass, A. J.; Tillett, J. G. In *The Chemistry of Sulphonic Acids, Esters and their Derivatives*; Patai, S.; Rappoport, Z. Eds.; Wiley: New York, 1991; pp. 789-878. (b) Roberts, D. W.; Williams, D. L. *Tetrahedron* **1987**, *43*, 1027-1062.
- (a) Metz, P.; Fleischer, M.; Fröhlich, R. *Tetrahedron* **1995**, *51*, 711-732. (b) Metz, P.; Fleischer, M.; Fröhlich, R. *Synlett* **1992**, 985-987.

3. A related intermediate featuring a carbanionic center in the endocyclic  $\beta$ -position to sulfur is probably formed in the final step of our synthesis of methyl nonactate, see: Metz, P.; Meiners, U.; Cramer, E.; Fröhlich, R.; Wibbeling, B. *Chem. Commun.* **1996**, 431-432.
4. Hahn, R. C.; Tompkins, J. *Tetrahedron Lett.* **1990**, 31, 937-940.
5. De Lima, C.; Julia, M.; Verpeaux, J.-N. *Synlett* **1992**, 133-134.
6. While racemic **2b** was used, enantiomerically pure **2a** was employed (ref. 2); optical rotation data for **7a**:  $[\alpha]_{\text{D}}^{20} = -59.1$  ( $c = 1.24$  in THF), for **6a**:  $[\alpha]_{\text{D}}^{20} = -120$  ( $c = 1.14$  in THF).
7. Fleming, I.; Patel, S. K.; Urch, C. J. *J. Chem. Soc., Perkin Trans. 1* **1989**, 115-124.
8. Sultone **2b** was completely recovered after lithiation and subsequent treatment with 10 equiv. of (chloromethyl)trimethylsilane (THF,  $-78$  °C to  $20$  °C). In the presence of triethylamine (10 equiv.), the chloride effected partial conversion to **7b**.
9. Kocienski, P. J. *Tetrahedron Lett.* **1979**, 2649-2650.
10. Simpkins, N. S. *Sulphones in Organic Synthesis*; Pergamon Press: Oxford, 1993.
11. Sultones **7a** and **7b** were obtained as single diastereomers, respectively. The stereochemical assignment for these compounds follows from diagnostic  $^1\text{H}$ ,  $^1\text{H}$  coupling constants and NOE difference data (cf. ref. 2a).
12. Rondstedt jr., C. S. *J. Am. Chem. Soc.* **1954**, 76, 1926-1929.
13. All 4 diastereomers of **8** resulting from the variable relative configuration at the acetal carbon and the carbon  $\alpha$  to sulfur were present in equal amounts.
14. Metz, P.; Stölting, J.; Läge, M.; Krebs, B. *Angew. Chem.* **1994**, 106, 2275-2276; *Angew. Chem. Int. Ed. Engl.* **1994**, 33, 2195-2197.
15. Cf. (a) Bäckvall, J.-E.; Byström, S. E.; Nordberg, R. E. *J. Org. Chem.* **1984**, 49, 4619-4631. (b) Bäckvall, J.-E.; Granberg, K. L.; Hopkins, R. B. *Acta Chem. Scand.* **1990**, 44, 492-499.
16. Kupchan, S. M.; Baxter, R. L.; Chiang, C.-K.; Gilmore, C. J.; Bryan, R. F. *J. Chem. Soc., Chem. Commun.* **1973**, 842-843.
17. Grieco, P. A.; Oguri, T.; Gilman, S. *J. Am. Chem. Soc.* **1980**, 102, 5886-5891.
18. (a) Roberts, M. R.; Schlessinger, R. H. *J. Am. Chem. Soc.* **1981**, 103, 724-725. (b) Schlessinger, R. H. In *Organic Synthesis Today and Tomorrow*; Trost, B. M.; Hutchinson, C. R. Eds.; Pergamon Press: Oxford, 1981; pp. 251-258.
19. Wakamatsu, T.; Miyachi, N.; Ozaki, F.; Shibasaki, M.; Ban, Y. *Tetrahedron Lett.* **1988**, 29, 3829-3832.
20. (a) Metz, P.; Cramer, E. *Tetrahedron Lett.* **1993**, 34, 6371-6374. (b) Metz, P.; Meiners, U.; Fröhlich, R.; Grehl, M. *J. Org. Chem.* **1994**, 59, 3687-3689.

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